Toluene induced hypokalaemia: case report and literature review

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Abstract
Generalised weakness is a common complaint. A case is presented of toluene induced hypokalaemia in a 22 year old woman who presented with generalised weakness. The effect of toluene and causes of weakness and hypokalaemia in this setting are discussed.

(Keywords: toluene; potassium; metabolic acidosis)

The complaint of generalised weakness is common in the practice of emergency medicine. Often, objective findings are few and the diagnosis elusive. When a patient with objective, generalised motor weakness or paralysis presents, the differential diagnosis is more limited. A variety of electrolyte abnormalities have been associated with weakness. These include hypokalaemia, hyperkalaemia, hypocalcaemia, hypocalcaemic tetany, hyperphosphataemia, and hypermagnesaemia.

Toluene has been a reported cause of hypokalaemia. We present a case of generalised weakness associated with hypokalaemia and toluene misuse.

Case report
A 22 year old Hispanic woman was transferred to our emergency department with the presumptive diagnosis of Guillain-Barré syndrome (GBS). Two days before admission she developed numbness and tingling in her thighs. Later that day, when she awoke from a nap, she was unable to get out of bed. She described muscle aches and severe weakness in all extremities. This persisted and worsened over the next two days until she was completely unable to care for herself. She required assistance to get out of bed and was unable to walk. She had control of her bowel and bladder function and was able to eat with assistance.

Two days before the onset of her symptoms, the patient was sniffing spray paint. She admitted to inhaling spray paint as often as every few weeks. She denies other drug misuse, but drinks alcohol.

Her past medical history was significant for a motor vehicle accident resulting in a head injury 14 years previously. She had a craniotomy at that time, but the details are unknown. She had no known neurological deficit as a result. Since then she has had a seizure disorder for which she takes phenytoin.

Upon arrival at our department, her temperature was 35.9°C, heart rate was 81 beats/min, respiratory rate was 20 breaths/min, and blood pressure was 144/72 mm Hg. She was alert with normal mental status and seemed unusually nonchalant about her illness. She had a small contusion over her right eyebrow. The neck was supple without pain or tenderness. Heart, lung, and abdominal examination were normal with the exception of mild obesity. She had moderate tenderness in all the muscle compartments. Vascular examination was normal. The neurological examination showed normal cranial nerves. Sensation and proprioception were intact. Deep tendon reflexes were unobtainable. Strength was decreased in the trunk and extremities. She was not able to lift her head. Lower extremity distal motor strength was slightly decreased at 4/5. Proximal muscles were 5/5, but limited because of muscle pain. Upper extremities were weaker with 2–3/5 strength proximally, and 3/5 distally. She had normal rectal tone.

With the history of drug misuse and the forehead contusion, a traumatic spinal cord injury was considered. Thus, the patient was placed in a cervical collar and radiographs of the cervical spine were obtained, which were negative. The urine analysis showed a specific gravity of 1.015, 1+ blood, nitrite negative, pH −6.5, trace protein, 2 to 5 red blood cells and 5 to 10 white blood cells per high power field. Urine drug screen was negative. Phenytoin level was less than 2.5 µg/ml. Serum sodium was 141 mEq/l, potassium was 1.4 mEq/l, chloride was 111 mEq/l, carbon dioxide was 23 mEq/l, blood urea nitrogen was 8 mg/dl, creatine was 0.6 mg/dl, magnesium was 2.2 mg/dl, phosphorous was 2.7 mg/dl, and calcium was slightly low at 8.1 mg/dl. Serum aspartate aminotransferase was 71 IU/l, creatinine kinase was 2460 IU/l (100% MM), and LDH was 259 IU/l. An arterial blood gas measurement was taken shortly after potassium replacement had begun. It showed a pH of 7.42, PCO₂ of 32 mm Hg, PO₂ of 102 mm Hg, HCO₃ of 21 mmol/l and base excess of −2.5 mmol/l on room air. Urine sodium was 53 mEq/l, urine potassium was 24.3 mEq/l, urine chloride was 143 mEq/l, and urine creatinine was 143.7 mg/dl. Serum aldosterone was 1 ng/dl. The electrocardiogram showed prolonged Q-T interval and ST depression consistent with hypokalaemia.

EMERGENCY DEPARTMENT AND HOSPITAL COURSE
Potassium replacement via oral and intravenous infusion was begun in the emergency department. In the next 36 hours she received 310 mEq potassium, primarily orally, and her symptoms were completely resolved. At the
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Discussion
This patient presented to us with a provisional diagnosis of GBS. GBS is an inflammatory peripheral neuropathy characterised clinically as an ascending paralysis or weakness of variable degrees progressing over several days to three weeks.1 The clinical presentation usually begins with distal paresthesias followed by leg weakness. The weakness progresses proximally over the course of several days, often involving facial, oropharyngeal or respiratory muscle weakness requiring mechanical ventilation. This is commonly accompanied by pain in the large muscles. Deep tendon reflexes are markedly diminished or absent and sensation is intact. The signs and symptoms in this patient mimicked GBS. The differential diagnosis to be considered in this clinical setting includes spinal cord compression, transverse myelitis, myasthenia gravis, polymyositis, metabolic myopathies, paraneoplastic neuropathy, hypophosphataemia, botulism, paralytic poliomyelitis, and tick paralysis.

This patient’s symptoms were attributable to profound hypokalaemia. Hypokalaemia is a well known, but uncommon cause of acute weakness.2 Muscle weakness is most prominent in the limbs with the respiratory cranial and bulbar usually spared. Deep tendon reflexes are usually diminished or absent. The differential diagnosis of hypokalaemia can be classified into four categories. These are inadequate intake, excessive gastrointestinal losses, intracellular shift, or renal causes. The renal causes include diuretics, mineralocorticoid excess (such as primary or secondary hyperaldosteronism, licorice ingestion, or Cushings’s syndrome), Liddle’s syndrome, or renal tubular acidosis. This is supported by metabolic acidosis. In doing so, she would have decreased glomerular filtration rate from decreased extracellular volume may cause mild prerenal failure. This would inhibit the kidney’s ability to excrete these organic acids, adding to an increased gap.5

The cause of the hypokalaemia is also uncertain. Possible explanations include increased mineralocorticoid because of volume contraction,13 increased urinary excretion because of the presence of poorly reabsorbed anions and low urinary chloride concentration as would be seen with urinary excretion of hippurate and benzoic acid accompanied by volume contraction, and high urine flow rate because of the osmotic load of hippurate.5 One possible explanation for the very low potassium associated with the normal acid/base status in our patient was the care she received before seeking medical care. Her friends had been bringing her food and presumably liquids while she was unable to walk. This might have allowed her to maintain her intravascular volume and urine flow rate. She would have been able to excrete the hippurate and benzoic acid and correct metabolic acidosis. In doing so, she would have further lowered her serum potassium by causing an intracellular shift and enhanced excretion. Her alcohol misuse and relative obesity may have increased the acidosis by inducing the cytochrome P-450 and by allowing a higher amount of toluene to be stored in the adipose tissue.15

TREATMENT
Treatment of toluene exposure lies primarily in the correction of the resulting metabolic disorders. Most patients will require parenteral hydration and attention to renal function. Recovery is usually rapid with the correction of
the electrolyte disorders.\textsuperscript{4, 7} Bicarbonate replacement may be needed in severe acidosis with a pH less than 7.2.\textsuperscript{7}