Nine cases of accidental exposure to dimethyl sulphate—a potential chemical weapon

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Abbreviation: DMS, dimethyl sulphate

Dimethyl sulphate (DMS) is an innocuous appearing, widely used, and highly toxic chemical. It is used both as a methylating agent in industrial chemical synthesis and in medical laboratories for chemical cleavage of DNA. It is readily absorbed through the skin, mucous membranes, and gastrointestinal tract. Delayed toxicity allows potentially fatal exposures to occur prior to development of any warning symptoms. Toxicity is manifested initially by mucosal inflammation of eyes, nose, oropharynx, and airways. This can progress to severe airway oedema and necrosis, and non-cardiogenic pulmonary oedema. Other systemic effects include convulsions, delirium, coma, and renal, hepatic, and cardiac failure. All these features make DMS a potential chemical weapon.

We report nine cases of varying degrees of inhalational exposure to DMS, occurring as a result of a single chemical spillage incident in the United Kingdom. Industrial poisoning is surprisingly rare and there are few previous reports in the literature outside China.

The exposure occurred in an industrial hose manufacture and repair plant in August 2003. Three male workers aged 28–32 years were in a 12×12×4 metre workshop wearing overalls, boots, and gloves, but no other protective gear. The workshop had open doorways leading into rooms containing another six people. The environmental temperature was 24.6°C outdoors and the workshop contained a working fan.

A short length of enclosed industrial hose had been sent from another company for repair. On opening the hose, about 125 ml of clear liquid spilled onto the concrete floor without making skin contact with the workers. Assuming the fluid was water the men initially ignored it. Two workers left the workshop for adjacent rooms after two and three minutes, respectively. The third man (patient 1) considered the possibility of a chemical spill and used a high pressure water hose to dilute the liquid.

Thirty to sixty minutes after exposure all three workers noted eye irritation with discomfort, lacrimation, and erythema. The supplying company was contacted and confirmed the hose had been filled with DMS. Patient 1 then poured sand onto the diluted DMS, sealed the contaminated sand in a bin, and washed the floor down again. He remained with the uncontained DMS for approximately one hour. The other workers remained in adjacent rooms during this period. All three briefly irrigated their eyes with saline.

Five hours after exposure, the three workers presented to hospital with DMS safety data sheets. All three had marked conjunctival injection, lacrimation, rhinorrhoea, and a burning foreign body ocular sensation; symptoms were most marked in the patient most exposed to DMS (patient 1).

Although further advice was sought from a national poisons information centre the men had their eyes irrigated with normal saline. Slit lamp examination revealed marked conjunctival injection and punctuate corneal erosions. These were most marked in patient 1, who also complained of clouding of his vision. Acuity testing was normal, and all patients were commenced on topical chloramphenicol.

All three were showered, put into hospital gowns, and admitted for observation for possible development of delayed non-cardiogenic pulmonary oedema.

At nine hours, patient 1 had developed increasing throat tightness, hoarseness, a persistent harsh cough like cough producing clear sputum, excessive salivation, chest tightness, and subjective dyspnoea. Patient 2 had developed epistaxis and also complained of a sore tight throat.

CASE REPORTS

Dmethyl sulphate (CH₃)₂SO₄ (DMS) is a methylating agent used industrially in the synthesis of pharmaceuticals, dyes, perfumes, and pesticides. It is also used medically for chemical cleavage of DNA sequences. DMS is highly toxic, corrosive, and has carcinogenic, mutagenic, and teratogenic potential. It is an oily, clear, colourless, or pale yellow fluid with a faint onion like odour. Its vapour density is 4.35 (air = 1). The melting point is −32°C; the boiling point is 188°C. At 20°C the vapour pressure is 0.5 mmHg.

Absorption occurs readily through the skin, mucous membranes, and gastrointestinal tract. After acute exposure, symptoms are usually delayed for several hours, allowing potentially fatal exposures to occur before the patient becomes aware of their plight. DMS hydrolyses slowly into sulphuric acid, methyl hydrogen sulphate, and methanol on contact with mucosal surfaces. Sulphuric acid and methyl hydrogen sulphate are highly irritative and corrosive to these surfaces. The methanol may be absorbed into the circulation and readily absorbed through the skin, mucous membranes, and gastrointestinal tract. After acute exposure, symptoms are usually delayed for several hours, allowing potentially fatal exposures to occur before the patient becomes aware of their plight. DMS is a potential chemical weapon.

The Germans first experimented with DMS as a potential chemical weapon in a recent large scale terrorist attack training exercise. The Germans first experimented with DMS as a potential chemical weapon during World War I because of its powerful vesicant properties.
A mild transient neutrophilia, lymphopenia, and monocytopenia were noted, as well as a transient isolated elevation in serum bilirubin to 49 μmol/l (normal range 2–17 μmol/l). Other blood tests, chest radiograph, and an electrocardiogram were all normal.

Intravenous dexamethasone 8 mg (0.1 mg/kg) eight hourly was commenced. Gas exchange remained satisfactory overnight and he was extubated the following day. His chest tightness had resolved, but rhinorrhoea persisted for 72 hours after extubation. A cough productive of clear sputum persisted for three weeks, occasionally accompanied by wheezing. His visual blurring and discomfort cleared by day four and the conjunctival injection resolved completely by two weeks. Patient 1 returned to his normal work and exercise regime four weeks later. On follow up at three months the patient was entirely asymptomatic.

Patient 2 had normal observations with no hoarseness or stridor on repeat examination. His epistaxis was minor and resolved spontaneously. His progress was uneventful and he was discharged the following day. He noted nasal stuffiness, mild throat discomfort, and increased salivation for three days. His eye discomfort settled within 48 hours, and the erythema resolving after two weeks.

Patient 3 was discharged after overnight observation. His eye discomfort and rhinorrhoea settled within 24 hours, and the conjunctival erythema resolving by one week.

Six people working in the building did not enter the room where the spill occurred. Five of these developed symptoms, four presenting to hospital the following day. They all had mild conjunctival injection; two had a few punctuate epithelial erosions. One also described slight nasal symptoms and another mild throat irritation.

DISCUSSION

The toxicity of DMS, coupled with its availability, innocuous appearance, and delayed onset of potentially fatal symptoms make it a potential chemical agent of terrorism of which emergency physicians should be aware.

These cases demonstrate the characteristic delay in symptom onset previously reported in the literature. The patients who were most exposed to DMS developed symptoms of toxicity more quickly, and their symptoms were worse and longer lasting. The mucous membrane injury initially developing in the eyes and then progressing to nasal and upper airway symptoms is characteristic. The raised white cell count is also characteristic. Fortunately none of our cases developed the severe laryngeal oedema, non-cardiogenic pulmonary oedema, toxic shock, encephalopathy, myocardial damage, or methanol toxicity that can occur with greater exposures.

General treatment of DMS toxicity involves minimising exposure of patients and decontamination, together with close observation and supportive management. If there are manifestations of neurotoxicity one should evaluate for and treat possible associated methanol toxicity. Early high dose steroids are thought to reduce lung toxicity, decreasing the incidence and severity of non-cardiogenic pulmonary oedema.5 11

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REFERENCES

12. Met Office data: FitzRoy Rd, Exeter, Devon, EX1 3PB, United Kingdom.